



$\frac{C_{NaOH}}{C_{H_nA}} \Rightarrow \frac{0.32}{1 \times 160} = \frac{80 \times 10^{-3} \times 0.05}{n} \Rightarrow n = 2$

231. كبريت 3

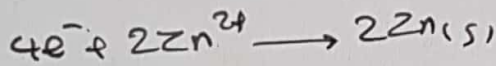
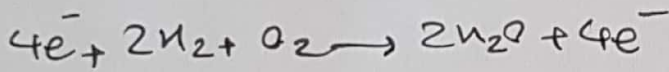
$[H^+] = \alpha \cdot C \Rightarrow 4 \times 10^{-3} = \alpha \times 0.1 \rightarrow \alpha = 4 \times 10^{-2} \rightarrow \% \alpha = 4$

$pH = -\log [H^+] = -\log(4 \times 10^{-3}) = -2.6028 = 2.4$

نزلتة - كوكبة - زور.

232. كبريت 1

233. كبريت 2



$\frac{26}{2} \times 3 = \frac{0.5}{1} \rightarrow \text{لح كاز} = 4.48 \times \frac{100}{20} = 5.6 L$

234. كبريت 4

235. كبريت 4